Pore-scale imaging of hydrogen displacement and trapping in porous media

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HIGHLIGHTS

- Hydrogen injectivity and recovery in rock is imaged with x-ray computed micro-CT.
- Hydrogen recovery decreases with increasing reservoir depth.
- Hydrogen recovery increases with increasing brine flow rate.
- Hydrogen trapping occurs via snap-off processes.
- Nitrogen is a poor proxy for hydrogen.

ABSTRACT

Hydrogen can act as an energy store to balance supply and demand in the renewable energy sector. Hydrogen storage in subsurface porous media could deliver high storage capacities but the volume of recoverable hydrogen is unknown. We imaged the displacement and capillary trapping of hydrogen by brine in a Clashach sandstone core at 2–7 MPa pore fluid pressure using X-ray computed microtomography. Hydrogen saturation obtained during drainage at capillary numbers of <10⁻⁷ was ~50% of the pore volume and independent of the pore fluid pressure. Hydrogen recovery during secondary imbibition at a capillary number of 2.4 × 10⁻⁶ systematically decreased with pressure, with 80%, 78% and 57% of the initial hydrogen recovered at 2, 5 and 7 MPa, respectively. Injection of brine at increasing capillary numbers up to 9.4 × 10⁻⁶ increased hydrogen recovery. Based on these results, we recommend more shallow, lower pressure sites for future hydrogen storage operations in porous media.

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**Introduction**

The storage of surplus electrical energy could increase the utility of renewable energy sources thereby reducing the environmental impact of energy generation [1,2]. Storage mediums like batteries, compressed air, capacitors, or fly-wheels are only applicable for the short-term and small-scale electricity storage [3]. The use of hydrogen (H₂) as an energy vector however could provide long-term storage to balance the intermittent demand and supply challenges affecting renewable methods [3]. Considering the low energy density of H₂ [4], storage of H₂ in large storage sites is the only solution for economical, gigawatt to terawatt-scale H₂ storage. Underground storage of H₂ in salt caverns is an established technology [5], yet, only subsurface porous media (saline aquifers and depleted oil and gas fields), can provide TWh storage capacities that could balance seasonal demands [6]. Relative to cavern storage, inter-seasonal storage in porous media is poorly understood and introduces a range of other complexities [2], including potential pore-clogging by microbial growth [6], gas loss out of diffuse lateral boundaries, viscous fingering of H₂ into reservoir brines, residual trapping and possible reactions with the primary formation fluid and reservoir minerals [7–9]. Each of these requires consideration of multiphase flow, transport and reaction [2,7].

Recent years have seen several studies on H₂ geochemistry [10–14] and the elucidation of many important flow characteristics of H₂ including: Hydrogen relative permeability and capillary pressure [15–18], interfacial tension (IFT) [19–21], and H₂/brine/rock contact angles [16,19,20,22–28]. Despite these advances important unknowns remain, e.g. the amount and type of cushion gas required in different geological settings, the risk for H₂ leakage via lateral boundaries and wells, the extent of microbial clogging and hydrogen consumption, and the degree of capillary trapping of H₂ as a function of pore size distribution.

Amongst the remaining unknowns, capillary trapping, which leads to a reduced volume of recoverable H₂ in this way impacting the economic feasibility of the H₂ storage operation, is of principal relevance [7]. Pore scale observations are particularly valuable for the assessment of capillary trapping [24]. Additional data describing the H₂ fluid flow and capillary trapping in different porous formations and under varying conditions are vital to make accurate predictions of the H₂ plume development and to define optimum production strategies [7]. The capillary forces that control capillary trapping also control the relative permeability [7]. As such studies on residual trapping give insight to a range of crucial input parameters for pore-scale and reservoir H₂ storage models.

The effects of the brine pore fluid pressures and of brine flow rate on capillary trapping of H₂ are hitherto unknown. The brine pore fluid pressure is linked to the reservoir depth via the hydrostatic gradient, hence an investigation of the latter could facilitate the definition of an optimal storage depth for gas recovery. The flowrate affects the capillary number, Nc, via equation (1):

\[
N_c = \frac{q \mu}{\sigma}
\]

where \(q\) is the flow, \(\mu\) is the viscosity of the invading fluid and \(\sigma\) is the interfacial tension (IFT). As \(N_c\) increases, viscous forces dominate capillary forces and capillary trapping of the non-wetting phase decreases [29]. This capillary desaturation with increasing \(N_c\) depends on the rock type [30] and on the kind of nonwetting phase [29] and is observed around critical capillary numbers of 10⁻⁵ to 10⁻⁸ [29].

A non-destructive standard technique for visualization and analysis of wetting and non-wetting phase displacement processes in porous media is x-ray micro-computed tomography (µCT) [31]. Pore-scale models are directly reliant on µCT images to build up the basic models for appropriate understanding of gas storage operations [32]. Previously published µCT studies on two-phase flow in porous media involving gas have largely focused on supercritical carbon dioxide (sCO₂) in sandstones and limestones. Reported saturations of sCO₂ in non-aged (i.e. not wettability altered by exposure to oil or organic acids) water-wet rocks range from 30 to 60% during drainage and from 15 to 30% during imbibition, at \(N_c\) between 10⁻⁸ to 10⁻⁶ and 10⁻⁸ to 2 × 10⁻⁵, respectively [31,33–36]. Studies using subcritical, gaseous carbon dioxide (CO₂), which could be more readily compared to H₂, are scarce. Yet, the characteristic trapping curves for sCO₂ and CO₂, which show the residual saturation \(S_{nu} (\%)\) as a function of initial saturation \(S_{init} (\%)\), are not significantly different [37]. Saturations of nitrogen (N₂) during drainage and imbibition in sandstones with 6–22% porosity are 43–64% and 43%, respectively, at \(N_c\) between 1–7x10⁻⁸ [38,39]. Carbon dioxide trapping increases at lower pressures due to decreasing water contact angles [40]. Unlike in CO₂ storage, trapping is not desirable in geological H₂ storage as it leads to unrecoverable H₂.

Some observational studies have suggested that H₂ is less wetting on rocks than CO₂ [23,25]. At the time of writing four µCT studies, one micro-model study and one nuclear magnetic resonance (NMR) study on H₂ exist: Al-Yaseri et al. (2022) [41] used NMR to find H₂ \(S_{nu} (\%)\) and \(S_{init} (\%)\) of 4% and <2%, respectively, in a Fontainebleau sandstone at 0.4 MPa and ambient temperature. Higgs et al. (2021) [19] µCT-imaged H₂ injection into a 5 mm diameter and 7.6 mm length core of Bentheimer sandstone at pressures of 6.8–20.8 MPa and documented decreasing IFT between H₂ and water with increasing pressure (72.5 mN m⁻¹ at 6.9 MPa to 69.4 mN m⁻¹ at 20.7 MPa). Jha et al. (2021) [42] conducted a single cycle H₂-brine displacement sequence in a 5 mm diameter and 15 mm length core of Gosford sandstone, and used µCT to calculate an \(S_{nu} (\%)\) of 65% during drainage and a \(S_{nu} (\%)\) of 41% after brine imbibition. Rock samples in Jha et al. (2021) [42] and Higgs et al. (2021) [19] were sufficiently short for capillary end effects to dominate the flow behaviour, where the wetting phase accumulates close to the production face of the core [43]. The experiment by Jha et al. (2021) [42] was performed at ambient pressure and leaves open the question of variations in H₂ wetting and flow behaviour at reservoir conditions [22]. Increasing pore fluid pressures have been reported to increase the gas saturation and pore network connectivity for N₂ in a brine wet, aged Bashijigike tight sandstone (0.6% gas saturation at 2 MPa versus 43% at 8 MPa) [38]. The effect of increases in \(N_c\) from 7.7 × 10⁻⁷ to 3.8 × 10⁻⁴ on the H₂ flow through a sandstone micro-model at 0.5 MPa was investigated by Lysyy
et al. (2022) [24] who found that $S_{\text{wri}}$ increased correspondingly from 18% to 79% [24]. Recently, Jangda et al. (2022) [44] reported $S_{\text{wri}}$ and $S_{\text{wur}}$ of 36% and 25%, respectively in unaged Bentheimer sandstone at 10 MPa and 50 °C [44]. In this work, we used μCT to investigate the displacement and capillary trapping of H$_2$ by brine in 4.7 mm diameter and 54–57 mm length Clashach sandstone cores as a function of gas and brine pore fluid pressure (2–7 MPa) and flow rate (20–80 μl min$^{-1}$, corresponding to H$_2$ and brine bulk $N_2$ of 1.7-6.8 × 10$^{-8}$ and 1.2-9.4 × 10$^{-8}$, respectively). The objectives of this work were to 1) deduce the effect of pore fluid pressure on H$_2$ capillary trapping, where the pore fluid pressure can be related to injection pressure and reservoir pressure in a H$_2$ storage scenario, allowing the definition of optimum storage depths for H$_2$ recovery, and 2) deduce the effect of brine flow rates on H$_2$ capillary trapping, to determine if increased recovery is possible with increased flow rate.

We show that the H$_2$ $S_{\text{wri}}$ is independent of pore fluid pressure but that higher brine pore fluid pressures during secondary imbibition increase capillary trapping of H$_2$, indicating decreased H$_2$ recovery at greater reservoir depths and hence less favourable conditions for H$_2$ storage. Higher flow rates during secondary imbibition on the other hand decrease capillary trapping, indicating that some trapped H$_2$ may be recovered under high flow conditions. Our work advances the current understanding of H$_2$ flow in porous media by delivering the first realistic estimates on how much H$_2$ can be recovered from the subsurface after injection, disregarding any H$_2$ loss by microbial or geochemical reactions, as a function of reservoir depth and flow rate.

**Materials and methods**

**Core-flooding experiments**

We performed unsteady and steady state, two-phase core flooding experiments with H$_2$ (purity 99.9995 vol%, BOC Ltd.) and brine (0.5 M CsCl or 2 M KI, Sigma-Aldrich) in a Clashach sandstone (Permian, aeolian sandstone from near Elgin in Scotland; the composition of Clashach is: ~96 wt% quartz, 2% K-feldspar, 1% calcite, 1% ankerite [45], porosity 11.1–14.4% [45–47]) at ambient temperature (~293 K). The Clashach sandstone, with its low mineralogical variability, yet still comparably big pore space, was chosen in order to facilitate the visualization of the injected H$_2$ and the interpretation of the results.

Two types of experiments were carried out: The first set of experiments (from now on referred to as UoE (University of Edinburgh) experiments) was carried out using the μCT instrument at the University of Edinburgh and were aimed at imaging the displacement and capillary trapping of H$_2$ by brine as a function of saturation after drainage and imbibition under different experimental conditions. Three UoE experiments were carried out that investigated the effect of pore fluid pressure (2–7 MPa), whereof each was repeated once (UoE exp. 1–3, Table 1). The standard error on the H$_2$ saturation in the repeated experiments was calculated as the standard deviation devied by the square root of the number of repeated experiments. One UoE experiment looked at the

![Table 1](https://doi.org/10.1016/j.ijhydene.2022.10.153)
effect of the H₂/brine injection ratio on the H₂ saturation (UoE exp. 4, Table 1). This experiment was carried out to evaluate the effect of H₂ injection into aquifers with different magnitudes of brine flow (hydrodynamic aquifers), and to simulate the far field situation, where H₂ and brine move together.

Another UoE experiment investigated the effect of secondary drainage and imbibition (UoE exp. 5, Table 1). UoE exp. 6 used N₂ instead of H₂. This experiment was undertaken in order to compare the flow behaviour of the two gases, which is of relevance because N₂ is sometimes used as an analogue for H₂, e.g. in permeability measurements [10,17]. Because rearrangement processes in the pore volume were noted previously for N₂ [38], UoE exp. 7 (Table 1) was undertaken to examine the stability of H₂ in the rock volume over an experimental duration (10 h): Ten PV of H₂ were injected into a brine-saturated rock while keeping the pressure constant inside the pressure vessel, with imaging undertaken at the start and after 10 h. The effect of bulk capillary number on drainage and imbibition was deduced by a comparison of the results from UoE exp. 2 and 5 which used flow rates of 20 and 80 μl min⁻¹, respectively, and 5 MPa injection pressure (Table 1).

The second type of experiment (from now on referred to as the ‘dry rock experiment’, Table 1) used synchrotron radiation (Diamond Light Source, I12-JEEP tomography beamline) to capture the time-resolved displacement of H₂ by brine in a dry, H₂-saturated rock. This experiment served as a base of comparison to the displacements of H₂ in wet, brine-saturated rock (UoE experiments). The chosen set of experiments allowed for the detailed assessment of the mechanisms behind H₂ and brine displacement processes in Clachash sandstone and allowed quantifying the hitherto unknown H₂ recovery from porous rock as a function of pressure/depth and flow.

All experiments used a bespoke x-ray transparent core holder for a 5 mm diameter rock core, which was a scaled-up version of the cell described by Fusseis et al. (2014) [48]. In UoE experiments a carbon-fibre reinforced PEEK pressure vessel was used to permit good x-ray transparency. The dry rock experiment used an aluminium pressure vessel to comply with the health and safety requirements at Diamond Light Source. The specifications of the pressure vessels of PEEK and aluminium were engineered with safety factors of 2 times or more over and above the maximum applied confining pressure (10 MPa; Table 1).

Rock cores for the experiments were obtained by diamond drill coring with a water-flushed chuck, followed by preparation of the core ends by grinding on a lathe. Experiments used a set of four high-pressure pumps (Cetoni Nemesys™, flow-rate range 0.072 nl s⁻¹ to 13.76 ml s⁻¹); One for the injection of H₂, one for the injection of brine, one to hold the backpressure and one for the confining pressure (Fig. 1). A bespoke manifold system composed of high-pressure 1/8” and 1/16” 316 stainless steel and 1/16” PEEK tubing (Swagelok, Top Industrie and Cole Farmer, respectively) connected the pumps to the core-flood cell (Fig. 1). Additional pressure transducers (ESI Technology; accuracy 0.1% full-scale) were coupled to the flow system at the inlet and outlet to allow for higher precision pressure monitoring than was possible using the integral pressure gauges in the syringe pumps. Cyclic H₂ and brine injections used a Clashach outcrop sample without further cleaning of 4.7 mm diameter and a relatively long length of 54—57 mm to avoid the influences of capillary end effects [49,50]. To prevent leakage of H₂ into the confining fluid, the rocks were jacketed in aluminium foil and polyolefin heat-shrink tubing and sealed with silicone adhesive between the conical-ended pistons within the pressure vessel. In UoE experiments, a water-wet rock was first saturated with brine (0.5 M CsCl) at a flow rate of 70 μl min⁻¹. Afterwards, H₂ was injected (drainage) into the brine-saturated rock at flow rates of 20—80 μl min⁻¹, based on desired capillary-regime NC of 1.7-6.8 × 10⁻⁶ (The viscosity of H₂ is 9.01 μPa s at 298K and 4.7 MPa [51] and the IFT between H₂ and water is 72.6 mN m⁻¹ at 298K and 5 MPa [52,53]). Subsequently, the brine was re-injected (imbibition) at flow rates of 20—80 μl min⁻¹, resulting in NC of 2.4-9.5 × 10⁻⁶ (using the same IFT between H₂ and water of 72.6 mN m⁻¹ at 5 MPAs and 298 K [52,53] and a viscosity of 1.247 × 10⁻³ Pa s at 5 MPa that was estimated from the reported 1.2503 × 10⁻³ Pa s and 1.233 × 10⁻³ Pa s at 0.1 MPa and 25 MPa, respectively, and 298 K [54]). The NC in the N₂ experiment was 3.5 × 10⁻⁸ during drainage (using an IFT of 73 mN m⁻¹ between N₂ and water at 298K and 10 MPa [37] and a viscosity of 1.89×10⁻³ Pa s at 5 MPa and 295K [51]). Each injection used ten pore volumes to ensure complete flushing of the sample cores with the injected fluid.

In the dry rock experiment, H₂ was directly injected into a dry rock at a flow rate of 70 μl min⁻¹. Subsequently, the brine (2 M KI) was injected at a flow rate of 5 μl min⁻¹, resulting in NC of 5 × 10⁻⁷ (using the same IFT between H₂ and water of 72.6 mN m⁻¹ at 5 MPa and 298 K [52,53] and a viscosity of 1.07×10⁻³ Pa s for 0.6 M KI and 293 K [45]).

The combined application of an x-ray transparent core holder and μCT allowed the visualization of the fluid saturation distributions at pore scale at each injection step. The difference in the x-ray attenuation coefficient of the fluids (H₂ and 0.5 M CsCl/2 M KI) provided an excellent contrast between the two fluid phases and the rock on the acquired μCT images, combined with the respective radiation energy in the two different laboratories.

3D volumes were acquired from the lower central portion of the sample to avoid the impact of capillary end effects on fluid saturation [49,50]. For the UoE experiments, image acquisition used a μCT instrument built in-house at the University of Edinburgh, comprising a Feinfocus 10—160 kV reflection source, a Micos UPR-160-air rotary table and a PerkinElmer XRD 0822 1 MP amorphous silicon flat panel detector with a terbium doped gadolinium oxysulphide scintillator. Data acquisition software was developed in-house. The following settings were used for UoE experiments: 120 keV, 16 W, 2 s exposure time, 1200 projections and 2 frames per stop. The voxel size was 5.4 μm³. In the dry rock experiment, time-resolved imaging of the H₂ and brine displacement processes was achieved by means of a 65 keV monochromatic beam detected by a high-resolution imaging camera with optical module 2 (PCO.edge 5.5, 7.91 × 7.9 μm/pixel with FoV 20 mm × 12 mm) using 17—25 ms exposure time and 900 projections. The voxel size was 7.9 μm³.
Tomographic reconstructions were undertaken by filtered back projection using Octopus 8.9 [55] on a GPU accelerated workstation. All subsequent image processing and analysis of tomographic data was performed using Avizo Version 9.1.1 (FEI, Oregon, USA). Data from UoE experiments were processed using a non-local means filter [56]. Processing of the dry rock experiment used a combination of median filter and unsharp mask to reduce image noise. Segmentation of UoE experiment data used a global threshold on the 2D greyscale image histogram, and encompassed two phases. In the water-wet scans, water and rock were treated as two discrete phases. In scans after brine and H₂ injections, the H₂ was treated as one phase and the brine and rock as a single separate phase, following protocols of Andrew et al. (2014) [34]. Holes and spots which were at the resolution limit of the data were removed from all datasets (applied thresholds corresponded to 33 and 53 voxels, respectively). Based on the segmented image of the water-wet scan in UoE experiments, a pore size distribution was calculated. The 3D image was separated into discrete pores using Avizo’s ‘separate objects’ module, which calculates a chamfer distance map of the pore-space and then applies a marker based watershed algorithm to the distance map to define discrete pore bodies as catchment basins, separated by the watershed which marks the location of pore throats (Figure A1d). Supporting information Fig. A1 shows the work flow for the water-wet scan.

In scans following brine and H₂ injections in UoE experiments, the segmented image was analysed in 3D using the ‘labeling’ and ‘label analysis’ modules to identify, label and measure the volume of each H₂ cluster. Hydrogen cluster size distributions were compared to the pore size distribution to evaluate the H₂ connectivity and identify trapping mechanisms during brine imbibition.

Capillary pressure

Recovered Clashach cores were submerged in 25% w/v NaOH solution (Fisher Scientific) for 2 h to remove the aluminium foil from the core surface, and rinsed in successive Milli-Q water, acetone and ethanol ultrasonic baths. Subsequently, the cores were cut and squared to the dimensions of 25 mm, overlapping the µCT visualized rock volumes, and cleaned ultrasonically with Milli-Q water to remove grinding products. Mercury injection capillary pressure (MICP) was performed on the cleaned cores, using a micromeritics automated mercury injection equipment (Autopore IV 9500) to estimate the capillary pressure-saturation relationship and pore size distribution. The pressure range tested was from vacuum to 379 MPa.

Results

Characterization of the pore space

The µCT-evaluated porosity of the Clashach sandstone from the segmented volume of the water-wet rock was 12.5%–13.5%, depending on the imaged region of the rock core. The MICP-evaluated pore throat size distribution showed a large number of very small pore throats with <5 μm radius and a small number of small to intermediate size pore throats (>5–90 μm) (Fig. 2). The largest pore throat had a radius of...
Hydrogen wetting behaviour and stability in UoE experiments

Hydrogen sat in the centre of the pore bodies. Residual brine sat in corners, pore throats (Fig. 3b and c) and, as a subtraction of the water saturated scan from the H₂-and brine filled rock revealed, in thin films around the grains (Fig. 3d). The injected H₂ remained stable within the pore volume under no-flow conditions and at constant pore fluid pressure over a time period of 10 h which was the maximum experimental duration (Fig. A2).

Effect of pore fluid pressure on hydrogen connectivity, saturation and recovery

Hydrogen saturation during drainage was independent of the pore fluid pressure with 49.8%, 51.7% and 39.7%–52.6% saturation at pore fluid pressures of 2, 5 and 7 MPa, respectively (Fig. 4a–c, f, Fig. A3). Hydrogen connectivity during drainage generally showed one large, connected cluster at all pore fluid pressures except for one out of three images at 7 MPa which showed three large clusters (Fig. 5, Fig. A3). During drainage, the largest H₂ cluster had a volume of $1 \times 10^6 \mu m^3$ at all pore fluid pressures except for the one run at 7 MPa with the disconnected clusters were the largest volume was $7 \times 10^7 \mu m^3$ (Fig. 5e). Hydrogen clusters during drainage were at all pore fluid pressures much larger than discrete pores with a maximum volume of $1.3 \times 10^6 \mu m^3$ (Fig. 5e, a). Comparing all H₂ cluster size distributions during drainage (Fig. 5f) reveals that all drainage curves, including two of the distributions at 7 MPa (squares and rhombi), have largely the same distribution, however one of the three distributions at 7 MPa (triangles) is distinct. This outlier experiment corresponds to the experiment showing a lower $S_{sw}$ (Fig. 4c).

Capillary trapping of H₂ during imbibition seemed independent of the pore fluid pressure with 10%, 12% and 4–21% of trapped H₂ at 2, 5 and 7 MPa, respectively (Fig. 4a–c, f, Fig. A3), corresponding to 20%, 22% and 11–43% of the initially injected H₂. During imbibition, large H₂ clusters were broken down into smaller clusters (Fig. 5a–d), in line with the visual changes of the H₂ clusters (Fig. 4a–c and f). The largest H₂ clusters after imbibition remained above the maximum pore size during all experiments except for one experiment at 7 MPa (Fig. 5g and a), showing that not only was H₂ trapped in discrete pore bodies but also as larger H₂ ganglia. The break-down of the largest H₂ clusters during imbibition caused the number of clusters in the size range log 4 to log 6 $\mu m^3$ to increase while the number of very small clusters of log 2–4 $\mu m^3$ typically decreased (Fig. 5b–d). Comparing all H₂ cluster size distributions during imbibition (Fig. 5g) shows that the distributions at 2 and 5 MPa are largely the same while the imbibition distributions at 7 MPa are distinct.

Injections of H₂ and brine into the same rock volume and at the same flow rates and pore fluid pressures of 2–5 MPa were repeatable with small standard errors between 0.01 and 0.66% (Fig. 4a and b, Fig. 5b and c). At 7 MPa very distinct $S_{sw}$ and $S_{nr}$ were measured (Fig. 4c and f). During drainage the standard error was 4.8% at an average H₂ saturation of 47.4%. The standard error during imbibition was 8.5% at an average saturation of 12.9%. The pressure differences between inlet and outlet during the experiments were within the error of the pressure sensors of 0.1% full-scale.

Effect of hydrogen/brine injection ratio on hydrogen connectivity and saturation

We studied the effect of the H₂/brine injection ratio on the H₂ saturation in order to evaluate the effect of H₂ injection into hydrodynamic aquifers, and to evaluate what happens in the far field, where fluids will be moving together. The results showed that the H₂ saturation and H₂ interconnected pore volume increased with increasing H₂/brine injection ratio from 32.6% at 4 $\mu l$ min $^{-1}$ H₂ plus 16 $\mu l$ min $^{-1}$ brine to 43.2% at 16 $\mu l$ min $^{-1}$ H₂ plus 4 $\mu l$ min $^{-1}$ brine (Fig. 6). The H₂ clusters in simultaneous injection experiments occupied many of the same pore spaces as the clusters after H₂ in the unsteady state experiments at the same pressure and total flowrate (Figs. 4b and 6, Fig. A6), and even at the lowest H₂ brine injection ratio of 4 $\mu l$ min $^{-1}$ H₂ plus 16 $\mu l$ min $^{-1}$ brine, the H₂ clusters were large, and spanning multiple pores (Fig. 6a). The percolation threshold, i.e. one connected path from inlet to outlet, was apparently only reached at 100% H₂ injection (Fig. 4b vs. Fig. 6, Fig. A6). The pressure differences between inlet and outlet during the simultaneous injection experiments were up to 0.05 MPa.

The H₂ cluster volume distributions were similar at different H₂/brine injection ratios (Fig. 7a). However, with increasing ratio the smallest H₂ clusters of volume $\log 2 \mu m^3$ decreased in number while the number of intermediate size $\log 2.5$ to $\log 6.25 \mu m^3$ H₂ clusters and the volume of the biggest cluster increased (Fig. 7a), confirming observations of
increasing H₂ saturation and connectivity with increasing injection ratio (Fig. 6).

**Effect of flowrate on hydrogen saturation and recovery**

At constant pore fluid pressure of 5 MPa, increases in the flowrate during drainage from 20 μl min⁻¹ to 80 μl min⁻¹, corresponding to bulk NC of 1.7 × 10⁻⁸ to 6.8 × 10⁻⁸, respectively, decreased the S_{nwi} from 51.7% to 47.7% (Fig. 4b and d). Correspondingly, increases in the brine flowrate during imbibition from 20 μl min⁻¹ to 80 μl min⁻¹, corresponding to bulk NC of 2.4 × 10⁻⁶ and 9.4 × 10⁻⁶, respectively, reduced the S_{nwr} from 11.5% to 7.2% (Fig. 4b and d).

In line with this, the H₂ cluster size distributions at the two flowrates showed that larger clusters were mobilized at 80 μl min⁻¹ (maximum cluster sizes of log 6.25 μm³ at 80 μl min⁻¹ vs. log 6.75 μm³ at 20 μl min⁻¹; Fig. 7b). At both flowrates the largest H₂ clusters were still bigger than the largest pore of <log 6 μm³ (Fig. 7b vs. Fig. 6a), indicating that H₂ was trapped also as larger ganglia.

**Secondary drainage and tertiary imbibition**

Secondary drainage and tertiary imbibition did not significantly change the H₂ saturation (47.9% and 7.0%) compared to primary drainage and secondary imbibition (47.6% and 7.3%), based on results at 5 MPa and 80 μl min⁻¹ flowrate (Fig. A4, Fig. 4d and e).

**Dry rock experiment**

In the dry-rock experiment, the brine entered the H₂-filled dry rock via piston-like displacement, (Fig. 8b), eventually recovering the H₂ entirely (Fig. 8d). Before all H₂ was recovered, there was an intermediate stage where previously brine-filled pores (Fig. 8b) showed several very small H₂ bubbles (Fig. 8d).

**Comparison to nitrogen**

The N₂ saturation was similar to the H₂ saturation during drainage (48.7% vs. 51.7 ± 0.66%, respectively) but N₂ saturation after brine imbibition was much higher (33.9% vs. 11.5 ± 0.64%, respectively; Fig. 4e, Fig. A5). The Nₐ of H₂ and N₂ during drainage was 1.7 × 10⁻⁸ and 3.5 × 10⁻⁸, respectively.

**Discussion**

**Pore space**

The μCT-evaluated porosity of the Clashach sandstone from the segmented volume of the water-wet rock of 12.5–13.6%
was within the range of the published porosities of 11.1–14.4% for Clashach sandstone [45–47]. The distributions of pore size and pore throat size as evaluated by μCT and MICP, respectively, suggested that mostly small pores of <50 μm radii were joined by very small throats of <5 μm radii with a few small to intermediate size throats of >5–90 μm radii in between (Fig. 2). The largest pore throat of 195 μm radius was probably measured at the surface of the rock core where the drilling process affected the pore space. The pore throat distribution for our Clashach sandstone sample was very similar to the pore throat distribution for Berea sandstone [57]. Compared to the pore throat distributions for Bentheimer sandstone and Doddington sandstone [57], our Clashach sandstone sample showed smaller pore throat sizes. Limestones generally show a wider pore throat size distributions than sandstones [57].

**H₂ flow behaviour and trapping mechanisms**

Hydrogen behaved as a non-wetting phase, filling the centre of the pores, with residual brine in the pore corners and throats (Fig. 3b and c), indicating a water wetting system. The largest H₂ cluster was much larger than discrete pores at any pore fluid pressure during drainage (Fig. 5), indicating a good connectivity of the H₂ [58]. Hydrogen trapping occurred via snap-
off of H₂ ganglia (Fig. 9). Snap-off competes with piston-like, i.e. pore-filling, displacement during the displacement of a non-wetting fluid by a wetting fluid in porous media. and is known as the swelling of water in the corner layers of a pore throat during water invasion in water-wet porous rocks until the threshold capillary pressure is exceeded, resulting in spontaneous filling of the throat with water and disconnection of the non-wetting phase which can lead to trapping [58,59]. Brine films around grains were not directly visible in the tomographic images (Fig. 3b and c) but were revealed by subtraction of the water-wet scan from the brine-saturated scan after H₂ injection, following the registration of the brine-saturated scan after H₂ injection to the water-wet scan (Fig. 3d and e). Fig. 3d and e suggest that brine films were discontinuous and very thin. When H₂ was injected into a dry rock, 100% of the injected H₂ could be recovered (Fig. 8e) which substantiated the theory that sub-resolution brine films around grains and snap-off of H₂ ganglia caused decreased H₂ recovery in experiments using an initially brine saturated rock (UoE experiments). The occurrence of several very small H₂ bubbles in the dry-rock experiment (Fig. 8d) indicated Roof snap-off [60] of H₂ ganglia.

**Effect of pore fluid pressure and hydrogen/brine injection ratio**

We observed no dependence of the H₂ saturation during drainage on pore fluid pressure, considering that 2 out of 3 experiments at 7 MPa showed the same H₂ saturation of ~50% as at 2 and 5 MPa (Fig. 4a–c,f, and Fig. A3). The one experiment

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at 7 MPa which had only 39.7% H₂ saturation (Fig. 4c, Fig. A3 and blue triangles in Fig. 5f) did also not have the same H₂ cluster size distribution as the remaining experiments (Fig. 5f), despite using the same experimental settings as for all other experiments at 7 MPa, and the log archives of the pore fluid pressures and injected volumes revealed no abnormalities. A shift in the distribution of cluster sizes can indicate a change in wettability, regardless of the measured H₂ saturation. Yet, as two of the results at 7 MPa showed a similar distribution as at the other pressures, it seems likely that this experiment is an outlier. The experiment was acquired after a filament change on the µCT apparatus, which implied that a slightly different part of the same rock core was imaged (13.6% vs. 12.5% porosity). Yet, in principle this should not have affected the results significantly, and subsequent experiments did return to show ~50% H₂ saturation, e.g. the H₂ stability experiment (Fig. A2).

The observed constant drainage H₂ saturations with increasing pore fluid pressures from 2 to 7 MPa are in line with a lack of a dependence of the H₂ wettability on pressure increases from 2 to 10 MPa in Berea and Bentheimer sandstone [22], with only very small increases of ~3°-6° in the H₂ contact angles at pressure increases from 2 to 7 MPa in Basalt [25], clay [28] and quartz [23], and with previous findings of no change in the characteristic trapping curves for CO₂ and N₂ at a wide range of pressure and temperature conditions [37]. The general anticipation of an increase in gas saturation with injection pressure [61,62] may still be valid over pressure ranges larger than the one investigated here. At unchanged wettability, the \( S_{\text{sur}} \) is controlled by the capillary pressure which in turn is controlled either by the fractional flow (during simultaneous injection) or by the viscous force pressure drop (during single fluid phase injection). Thus, the independence of the \( S_{\text{sur}} \) from the pore fluid pressure indicates that the viscous pressure drop...
is not being significantly altered by the changing pressure, e.g., the H₂ viscosity change is not having a major impact on the force required to drive flow. Significant loss of H₂ from the gas phase at higher pressures by dissolution into the brine is precluded by the low solubility of H₂ of ~0.02 mol kg⁻¹ at 2.5 MPa \cite{63}. The H₂ stability experiments showed that H₂ saturation at 5 MPa did not change over a time period of 10 h (49.55% at time zero vs. 49.53% 10 h after; Fig. A2), evidencing a stable result and no H₂ loss by dissolution. Recent reports of a significantly increased \( S_{\text{wr}} \) when using H₂-equilibrated brine over non H₂-equilibrated brine in H₂ and brine displacement experiments in Bentheimer sandstone at 10 MPa and 50 °C \cite{44}, indicate that employing a combination of high temperature and pressure causes significant dissolution of H₂.

Looking at the \( S_{\text{wr}} \) data only (Fig. 4a–c,f), there was no clear dependency on pore fluid pressure during imbibition. Any change may, however, have been masked by the high variation between the two results at 7 MPa (4–21% Fig. 4c and f). Considering that the experiment at 7 MPa with the low \( S_{\text{ori}} \) of 40% was identified as an outlier (see the discussion above \textit{4.3} and Fig. 5f, blue triangles) and that the \( S_{\text{wr}} \) is a function of the \( S_{\text{ori}} \) where a lower \( S_{\text{ori}} \) will tend to overestimate recovery \cite{57,64}, we may disregard the \( S_{\text{wr}} \) of 4%. The large increase in trapped H₂ in the second full primary drainage and secondary imbibition experiment at 7 MPa (Fig. 4f) was probably due to the poorer initial H₂ connectivity \cite{64}. The H₂ cluster size distribution for this experiment showed an increase in the number of intermediate size clusters compared to the other experiments (Fig. 5g, dark yellow squares), without however shifting the cluster distribution, suggesting no wettability change. This indicates that variation in one or more of the other thermophysical properties – density, viscosity, or IFT – have resulted in the impact on the pore scale fluid configuration (Fig. 10). Meanwhile, a poorer initial H₂ connectivity was not confirmed by the third repetition of primary drainage at 7 MPa (Fig. A3g). More experiments at 7 MPa are needed to confirm the result of increased trapping at higher pore fluid pressures.

Given a hydrostatic gradient of ~10 MPa/km an increase of the \( S_{\text{wr}} \) with increasing pore fluid pressure, as suggested by

![Fig. 8 – The dry-rock experiment summarized. (a) H₂-saturated, dry rock, (b) scan during brine imbibition shortly after appearance of the first brine in the rock, (c) scan during brine imbibition after 115 min, with several small, isolated H₂ bubbles inside pore bodies (d), and (e) brine-saturated rock after 2 h of imbibition showing 100% recoverability of H₂.](image-url)
the second experiment at 7 MPa (Fig. 4f), would indicate that deeper aquifers are less favourable for H₂ storage operations. However, unlike our unsteady state experiments which showed barely any pressure difference between inlet and outlet, in a real H₂ storage operation, the well pressure is significantly higher than the reservoir pressure and the brine is not injected, but naturally flowing into previously H₂-saturated rock when the H₂ is recovered again, due to a pressure difference. Our displacement study results are hence applicable to the fringe of the H₂-saturated zone, only, where pressure differences are very small. A decreased H₂ recovery with depth would not align well with other criteria for an economical and safe H₂ storage operation, such as a lower cushion gas requirement with elevated depth [5] and reduced risks for H₂-linked microbial activity at higher depths due to higher temperatures [6].

The pressure/depth effect on our results can be further related to the variation of key pore-scale displacement parameters for H₂ with depth (Fig. 10): The IFT of H₂ reduces with depth whereas the H₂ viscosity increases, both of which in theory should augment the displacement of H₂ with water and reduce the $S_{\text{wtr}}$ [66, 67]. However, increased $S_{\text{wtr}}$ with decreasing IFT and increasing viscosity has also been demonstrated [68] which may be due to a simultaneously increased likelihood of unstable displacement/non-uniform fronts [69, 70] during both drainage and imbibition processes at decreasing IFT and increasing viscosity, based on the augmenting effect of both parameters on $N_C$ (Eq. (1)). Such unstable displacement processes at higher pore fluid pressures, addition to a variation in the thickness of the brine thin films with pressure, may explain the decrease in interconnected H₂-filled pore volume from one large cluster at 2–5 MPa to three clusters at 7 MPa and have lead to increased snap-off and trapping during imbibition (Fig. 4f).

The $S_{\text{wtr}}$ of 10–21% was significantly lower than the previously reported H₂ $S_{\text{wtr}}$ of 41% for a Gosford sandstone under ambient conditions [42] but in line with 20–25% H₂ $S_{\text{wtr}}$ in a Bentheimer sandstone at 10 MPa and 50 °C [44]. As mentioned previously, the short length rock sample in Jha et al. (2021) [42] suggests that their results were affected by capillary end effects [49]. However, the relatively high bulk $N_C$ of 2.3–2.4 × 10⁻⁶ during brine imbibition in our experiments and those of Jangda et al. (2022) [44] may have mobilized more residual H₂ than under strict capillary regime conditions. Our results are higher than previously reported $S_{\text{wtr}}$ of <2% in a Fontainebleau sandstone at 0.4 MPa, ambient temperature and bulk $N_C$ of 3.5×10⁻⁸ [41], however the $S_{\text{wtr}}$ in this study was extremely low (4%).

Fig. 9 – Example of a snap-off event. (a) Labelled H₂-filled volume after drainage (orange), spanning over several pores and showing one large interconnected H₂-filled pore volume, and total pore space (transparent blue). (b) Labelled H₂-filled volume after brine imbibition (different coloured shades) and the total pore space (transparent blue) showing several, not connected H₂ ganglia and a snapped-off H₂ droplet (cobalt blue) in the centre, left hand side. (c) Pore body visualization of the same volume. (For interpretation of the references to color/colour in this figure legend, the reader is referred to the Web version of this article.)

Fig. 10 – Density (green line), viscosity (red line), and interfacial tension (blue line) of hydrogen versus depth (Hassanpouryouzband et al., 2021) [65]. (For interpretation of the references to color/colour in this figure legend, the reader is referred to the Web version of this article.)
During simultaneous injection of H₂ and brine, which may be representative of H₂ injection into hydrodynamic aquifers or simulate the far field conditions, H₂ saturation and H₂ interconnected pore volume increased with increasing H₂/brine injection ratio (Fig. 6). This indicated that a lower brine flow is favoured over high flow environments in terms of optimising the H₂ storage operation. The structure was apparently not percolating in any of the simultaneous injection experiments as opposed to during 100% H₂ injection (Fig. 6 vs. Fig. 4b), yet considering significant pressure differences of up to 0.05 MPa between inlet and outlet in simultaneous injection experiments which were not observed in experiments injecting 100% H₂, the connections between the H₂ clusters may have been broken when the injection (and thereby the pressure gradient) was stopped for the scan.

Effect of capillary number on initial and residual saturation

Classical pore-scale displacement theory predicts little change in residual phase saturation in response to increases in flow rate until the NC exceeds \(10^{-6}-10^{-5}\). However, for most subsurface reservoirs there will be rapid decreases at NC of \(10^{-4}\) or more, when viscous forces become dominant [57]. The bulk brine NC applied in this study (2.4-9.4 \(\times 10^{-5}\)) was within the range of little saturation change but exceeded the threshold of NC < \(10^{-6}\) for which the flow generally is said to be capillary dominated [57]. This may indicate that viscous forces caused a significant effect of NC on the Snwi in our experiments (Fig. 4b and d), and these forces are likely to be even greater at local scale than at bulk [57]. The H₂ cluster size distribution after imbibition at NC = 9.4 \(\times 10^{-6}\) was shifted with respect to the distribution at NC = 2.4 \(\times 10^{-8}\) (Fig. 7b), indicating a change in the wetting behaviour and supporting previous findings of preferential desaturation of larger clusters at higher NC [71].

We observed a 4% decrease in Snwi in our experiments when bulk NC was increased from 1.7 \(\times 10^{-8}\) to 6.8 \(\times 10^{-8}\) (Fig. 4b and d). Critical nonwetting phase NC of 2 \(\times 10^{-8}\) and \(10^{-5}\) during imbibition have been reported for water–gas systems and water-oil systems, respectively [72], indicating that the threshold of NC < \(10^{-6}\) for capillary dominated flow [57] is not rigid. However, considering reported increases in the H₂ saturation after drainage with increasing NC from 7.7 \(\times 10^{-7}\) to 7.7 \(\times 10^{-5}\) [24], and acknowledging the small observed difference in Snwi, we cannot conclude that the effect of flow rate was down to experimental variability.

Comparison to nitrogen

The N₂ saturation was comparable to the H₂ saturation during drainage at similar NC of 1-3x10^{-8} but the Snwi after imbibition was ~20% higher for N₂ than for H₂ (Fig. 4b and e). Using N₂ as a proxy for H₂ in experimental drainage and imbibition studies is hence not advisable. Considering the high degree of N₂ trapping, the use of N₂ as a cushion gas for H₂ storage operations which could reduce operational costs [7] seems favourable. Our results are lower than a reported 64% N₂ Snwi after drainage and 43% N₂ Snwi in a Berea sandstone (20-22% porosity) at 5.5 MPa and 20 °C [39], and higher than 43% N₂ Snwi in a Bashiğiçe tight sandstone (5.6% porosity) at 8 MPa and ambient temperature [38]. The trend in the differences of the N₂ saturation in the above studies follows the same trend as the differences in the porosities of the studied sandstones, with the Clashach sandstone (14% porosity) being intermediate between the two other rocks. This indicates that porosity differences between the different rock types applied in the above experiments defined the observed N₂ saturations, yet differences in the pore throats dimensions may equally have contributed or caused this. It also suggests that Snwi and Snwir depend strongly on (the local conditions within) each rock, and that these rock type/local effects may mask any effect of injection conditions, whereas trends in the rock-specific behaviour will be controlled by pressure and flow conditions. Meanwhile observations of N₂ Snwi and Snwir of 15–26% and 8–17%, respectively, in a Fontainebleau sandstone with 9.7% porosity at 0.4 MPa, ambient temperature and Nc of 3.5 \(\times 10^{-8}\) to 7 \(\times 10^{-7}\) [41] do not confirm the relation between initial and residual saturations and porosity. This suggests that other parameters such as the absolute permeability of a rock also shape the rock specific response to N₂ and brine displacement processes. More studies on different types of rock and under similar injection conditions are needed to better understand the rock-specific differences in Snwi and Snwir.

Suitability of the Clashach sandstone for hydrogen storage

It has been postulated that the low viscosity of H₂ will cause the gas to travel swiftly, making it unsuitable for displacing brine [69] and causing low H₂ injectivity. In this work, we showed that from an injectivity and recovery perspective, untreated Clashach sandstone is suitable for underground H₂ storage. However, sandstones aged by exposure to humic acids may be more suitable analogue rocks for experimental investigations of H₂ storage in porous media [23]. Considering that aging has previously been shown to alter the wettability of H₂ brine-quartz systems from highly water-wet toward intermediate-wet [23] the stated H₂ saturations for our untreated outcrop Clashach sandstone are expected to increase during drainage and decrease during imbibition, further the increasing suitability for H₂ storage.

Conclusion

In this work, a prima facie examination of H₂ flow and displacement processes in porous rock was carried out as a function of capillary numbers of 1.2-6.8 \(\times 10^{-8}\) for H₂ and 2.4-9.5 \(\times 10^{-6}\) for brine, and of pore fluid pressures between 2–7 MPa. Results showed no clear influence of pore fluid pressure on H₂ saturation during drainage in the investigated pressure range, with ~50% of the pore space saturated with H₂ during drainage at all pressures. During imbibition, 20%, 22% and 43% of the initially injected H₂ was trapped at 2, 5 and 7 MPa, respectively, and a capillary number of 2.4 \(\times 10^{-6}\), indicating that higher pressure, i.e. deeper reservoirs are less favourable for H₂ storage. Injection of brine at higher capillary numbers reduced capillary trapping and increased H₂ recovery. Hydrogen recovery was distinct from N₂ recovery, suggesting that N₂ is a poor proxy for H₂. Based on these results, we...
recommend more shallow, lower pressure sites for future H₂ storage operations in porous media.

Future work should aim to measure the influence of pressures above 7 MPa, elevated temperatures and rock aging on H₂ and brine displacement processes at a range of different capillary numbers and in different porous formations. Closer characterization of the pore space in terms of connectivity (Euler number), absolute and relative permeability, tortuosity and pore size could give further insight to understanding rock-specific responses during drainage and imbibition. Dynamic experiments using synchrotron light sources are recommended to examine displacement mechanisms closely. The presented experimental setup is suitable for investigating gas and brine displacement processes in a variety of porous samples, and is only limited by size of the pore matrix due to the resolution requirements of the μ-CT scanner.

**Author contributions**

The experimental kit was built by Butler and Thaysen. Experiments were carried out by Thaysen, Butler, Hassanpouryouzband, Freitas. Alvarez-Borges and Atwood were the supporting beamline scientists during the experiments at Diamond Light Source. Image reconstruction and tomographic analysis was carried out by Thaysen. Interpretation of the results was by Thaysen, Butler, Hassanpouryouzband, Krevor and Heinemann. The manuscript was written by Thaysen with review by Butler, Krevor, Heinemann, Hassanpouryouzband, Freitas and Edlmann. All authors have given approval to the final version of the manuscript.

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**Declaration of competing interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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**Appendix A. Supplementary data**

Supplementary data to this article can be found online at https://doi.org/10.1016/j.ijhydene.2022.10.153.

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